

AMINO ACID SUBSTRATE SPECIFICITY OF ASPARAGINYL-, ASPARTYL- AND GLUTAMINYL-*t*RNA SYNTHETASE ISOLATED FROM HIGHER PLANTS

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Abstract—AspNH₂-, Asp- and GluNH₂-*t*RNA synthetases were purified from *Phaseolus aureus*, their optimum assay conditions, substrate specificities and salt sensitivities were investigated AspNH₂-*t*RNA synthetase from β -cyanoalanine-producing (*Vicia sativa*), and non-producing (*P. aureus* and *V. faba*) species was able to utilize the analogue as a substrate irrespective of the source of the enzyme Asp-*t*RNA synthetase from *P. aureus* was able to utilize α -aminomalonate and *threo*- β -hydroxyAsp as a substrate The transfer of ¹⁴C-GluNH₂ to *t*RNA, catalyzed by GluNH₂-*t*RNA synthetase, was only inhibited by high concentrations of those analogues tested, albizzine was the most efficient, but no difference could be demonstrated between the substrate specificities of the enzyme isolated from an albizzine-producer (*A. julibrissin*) and a non-producer (*P. aureus*) species

INTRODUCTION

ASPARAGINYL (AspNH₂)-, aspartyl (Asp)- and glutamyl (GluNH₂)-*t*RNA synthetases have not previously been investigated in higher plants probably due to the relatively low levels of activity determined in studies on aminoacyl-*t*RNA synthetases¹

AspNH₂-*t*RNA synthetase has been purified from *Lactobacillus arabinosus*,² *Escherichia coli*³ and rabbit liver⁴ The enzyme has been shown to synthesize a specific form of AspNH₂-*t*RNA which acts as a repressor of AspNH₂-synthetase in experimental tumours⁵

Asp-*t*RNA synthetase has been purified from *L. arabinosus*⁶ and baker's yeast^{7,8} The occurrence of multiple forms of the enzyme has been reported in *Neurospora crassa*⁹ and *Bacillus brevis*.¹⁰

GluNH₂-*t*RNA synthetase has been purified from *E. coli*,¹¹ yeast, pig liver¹² and rat liver¹³ The enzyme is unusual, in that it can only be assayed by the ATP-PPi technique

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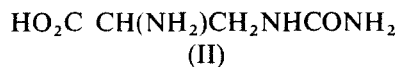
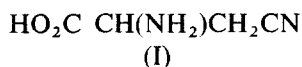
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when *t*RNA is present in the assay mixture¹¹⁻¹³ Certain species of *Bacillus* may lack a specific GluNH₂-*t*RNA synthetase. Glu-*t*RNA^{GluNH₂} has been shown to be formed by Glu-*t*RNA synthetase, which then may be amidated by a GluNH₂- or AspNH₂-dependent amidotransferase to form GluNH₂-*t*RNA^{GluNH₂}¹⁴

Investigations in the authors' laboratories have provided information about the substrate specificity of a number of higher plant aminoacyl-*t*RNA synthetases.¹ We have now purified the above synthetases and determined their general properties, because plants produce a number of non-protein amino acids, which might act as analogue substrates,¹⁵ our studies have placed particular emphasis upon the amino acid substrate specificity of the enzymes. The specificities of enzymes from a plant (*Phaseolus aureus*) failing to produce such analogues were compared with those of enzymes isolated from plants producing analogue molecules, e.g. AspNH₂-*t*RNA synthetase from *Vicia sativa* (a producer of β-cyanoalanine, I)¹⁶ and GluNH₂-*t*RNA synthetase from *Albizzia julibrissin* (which produces albizzine, II)¹⁷



RESULTS

Asparaginyl-tRNA Synthetase

Enzyme Fractionation

AspNH₂-*t*RNA synthetase was extracted and purified from 3 plant sources according to the scheme outlined in Table 1. Glycerol (15%, w/v) and mercaptoethanol (25 mM) were present routinely in the extraction buffer, PVP (MW ca 44 000, 0.3%, w/v) was only required when extraction was from leaf material, or from *V. sativa* seeds which contained high amounts of tannins. The precipitation of the enzyme in late ammonium sulphate fractions (60–70% saturation) allowed the removal of a large amount of protein in the earlier fractions. The endogenous level of synthetase activity (measured by ATP-PP_i exchange) of this fraction was high, but this was considerably reduced by acetone fractionation prior to the preparation being applied to the DEAE-cellulose column.

The fractionation scheme employed for the enzyme was essentially the same for each species studied, and each of the three enzymes eluted as a single peak of activity from the DEAE-cellulose column between 110 and 130 ml of gradient buffer. In the heterologous acylation experiments it was necessary only to pass a 60–70% saturated ammonium sulphate fraction through a column of Sephadex G75, to remove ribonucleases, *t*RNA and amino acids.

Optimum Assay Conditions

ATP-PP_i exchange assay The *K_m* for AspNH₂ was calculated as 1.8×10^{-4} M, a concentration of 5×10^{-3} M was routinely used to determine the *V_{max}* for particular enzyme fractions. The ATP and ³²PP_i concentrations were maintained at 4 mM, and the optimum Mg²⁺ concentration was determined as 2×10^{-2} M with a *K_m* of 5.8×10^{-3} M. The pH optimum was 7.8, with Tris-chloride and Tris-acetate being equally effective in promoting

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maximum exchange during 20 min at 40°. Putrescine, spermine and spermidine could not substitute for Mg^{2+} , but Mn^{2+} and Co^{2+} stimulated ATP-PP_i exchange at 30 and 17%, respectively, of the maximum rate achieved with Mg^{2+} .

TABLE 1 PURIFICATION OF AspNH₂-tRNA SYNTHETASE FROM THREE PLANT SPECIES

Species	Purification treatment	Total protein (mg)	Specific activity*	Relative purification	Contaminating synthetases†
<i>P. aureus</i> seed	Homogenate‡	1300	3.4	1	3200
	Ammonium sulphate fraction	192	21.6	6.3	965
	Acetone fraction	42.6	80.2	23.6	243
	DEAE-cellulose	5.2	566	166	18
<i>V. faba</i> seed	Homogenate‡	1762	2.8	1	2200
	Ammonium sulphate fraction	225	19.4	6.9	765
	Acetone fraction	47.8	69.3	24.7	192
	DEAE-cellulose	5.9	472	175	21
<i>V. sativa</i> seed	Homogenate‡	972	1.9	1	2350
	Ammonium sulphate fraction	142	12.1	6.4	872
	Acetone fraction	37.1	40.1	21.2	176
	DEAE-cellulose	4.1	287	151	24

* Specific activities are expressed as nmol PP_i exchanged/min/mg protein

† Expressed as % of AspNH₂-tRNA synthetase in each fraction

‡ Determined after removal of endogenous amino acids by Sephadex G25

tRNA esterification assay The K_m for AspNH₂ was calculated as 1.2×10^{-4} M, but routinely 1 μ Ci of undiluted ³H-AspNH₂ (10 nmol) was used for all assays (final vol. 1 ml). The optimum tRNA concentration was 0.2 mg/ml; concentrations higher than 1.0 mg/ml were inhibitory. The total level of aminoacylation reached a maximum after 12 min incubation at 37° and remained constant for 60 min. The ATP concentration was maintained at 10^{-3} M and the optimum Mg^{2+} concentration determined as 1.2×10^{-2} M with a K_m of 2.8×10^{-3} M. The pH optimum was determined as 7.8, with a sharp drop in activity above pH 8.0. Mg^{2+} could be replaced by various divalent cations at 10^{-2} M, activities for Mn^{2+} , Co^{2+} , Ni^{2+} and Ca^{2+} were 72, 45, 15 and 5% respectively of the values obtained with Mg^{2+} . Cd^{2+} , Zn^{2+} , Sn^{2+} , Sr^{2+} , Ba^{2+} , and the three polyamines were unable to catalyze the formation of AspNH₂-tRNA. Cr^{2+} appeared to catalyze a greater formation of ³H-AspNH₂-tRNA than Mg^{2+} , but this reaction was independent of ATP and was presumed to be due to the formation of a Cr^{2+} -³H-AspNH₂ complex.

Enzyme Stability

The enzyme could be stored in 40% (w/v) glycerol at -20° for 2 months with less than 10% loss of activity. Storage was possible also at 4° in 15% (w/v) glycerol for 48 hr without loss of activity, mercaptoethanol was unnecessary for either of these stabilizations. Half the enzyme activity was lost by heating in 0.1 M Tris-chloride buffer pH 7.8 and 15% (w/v) glycerol at 50° for 5 min. The addition of 4 μ mol ATP increased the time for 50% loss to 14 min, whilst simultaneous addition of AspNH₂ (50 μ mol) and ATP increased this time

to 27 min. Concentrations of urea above 4 M inactivated the enzyme during incubation at 37° for 5 min, but 70% of the activity could be recovered by dialysis of the extract against 0.1 M Tris-chloride buffer pH 7.8 containing 15% (w/v) glycerol and 25 mM mercapto-ethanol. *p*-Chloromercuribenzoate (*p*-CMB) inhibited the enzyme (DEAE-cellulose preparation containing 50 µg protein/ml) with a K_i of 5×10^{-5} M. Sodium azide at a concentration of 10^{-2} M did not inhibit the enzyme during the assay period.

Salt Effects

The enzyme when assayed by ATP-PP_i exchange was relatively insensitive to the addition of monovalent ions. The concentrations required to give 50% inhibition (50% I), were calculated for Li⁺, NH₄⁺, Na⁺, K⁺ and Cs⁺ chlorides as 0.7, 0.9, 2.0, 2.5 and 1.3 M, respectively, and values for CNS⁻, I⁻, Br⁻, Cl⁻, NO₃⁻ and MeCOO⁻ were 0.4, 0.5, 1.2, 2.5, 1.0 and 1.3 M respectively.

Aminoacylation of *t*RNA was much more sensitive to the presence of ions, although the relative values of 50% I for the various ions were little changed from those observed in the ATP-PP_i exchange assay. Plots of activity against ionic strength for NaCl and KCl gave markedly sigmoid curves with a sharp fall in activity between 0.15 and 0.25 M and 0.2 and 0.3 M, respectively. Plots of the same data in Debye-Hückel form (\log_{10} activity against $\sqrt{\mu}/1 + \sqrt{\mu}$) gave parallel straight lines for values of $\sqrt{\mu}/1 + \sqrt{\mu}$ between 0.28 and 0.4. Ammonium chloride inhibited aminoacylation at concentrations above 0.3 M, but increased the rate by 30% at 0.1 M. Tris was also inhibitory at concentrations above 0.3 M but had no stimulatory effect at lower concentrations. Ethanol (5%, v/v) gave a slight stimulation of activity at low *t*RNA concentrations by apparently lowering the K_m values for the *t*RNA.

TABLE 2 KINETIC PARAMETERS OF THE ATP-PP_i EXCHANGE STIMULATED BY AspNH₂ AND SEVERAL ANALOGUES USING AspNH₂-*t*RNA SYNTHETASE FROM *Phaseolus aureus*

Substrate	V_{\max}	Concn for V_{\max} (M)	K_m
Asparagine	100	5×10^{-3}	1.8×10^{-4}
Isoasparagine	96	5×10^{-3}	2.6×10^{-4}
Aspartate- β -hydroxamate	81	8×10^{-3}	7.8×10^{-4}
β -Cyanoalanine	36	5×10^{-2}	5.8×10^{-3}
<i>threo</i> - β -Hydroxyasparagine	9	*	4.2×10^{-2}
α,γ -Diaminobutyric acid	7	*	5.6×10^{-2}

Results are expressed with respect to the L-form of the amino acid, K_m values are expressed as molar concentrations, V_{\max} expressed as a % of the V_{\max} for AspNH₂.

2-Amino-2-carboxyethanesulphonamide, *erythro*- β -hydroxyAspNH₂, *O*-phosphoserine, *meso*-diaminosuccinic acid monoamide, 5-diazo-4-oxonorvaline, 5-bromo-4-oxonorvaline, Asp- β -hydrazide, 5-chloro-4-oxonorvaline, β -methylAspNH₂ and *N*- α -methylAspNH₂ failed to promote ATP-PP_i exchange at concentrations of 5×10^{-2} M.

* Values extrapolated from Lineweaver-Burk plots.

Amino Acid Substrate Specificity

ATP-PP_i exchange reaction The ability of AspNH₂-*t*RNA synthetase from *P. aureus* to utilize various analogues is shown in Table 2. Lineweaver-Burk plots for all analogues

tested were linear. None of the analogues promoting ATP-PP_i exchange increased the level of exchange when optimum concentrations of AspNH₂ were present. Only aspartic acid- β -hydroxamate inhibited AspNH₂-dependent exchange, when added at a concentration 50-fold in excess of AspNH₂.

The hydroxamate itself exhibited abnormal substrate kinetics, the rate of activation progressively decreasing at concentrations above 10^{-2} M, probably due to the nucleophilic inhibition of pyrophosphorolysis by hydroxamates.

Aminoacylation of tRNA Transfer of an analogue to tRNA was indirectly assessed by measuring the inhibition of ³H-AspNH₂ transfer to tRNA produced by various concentrations of unlabelled analogues (Table 3), tRNA was always the limiting component in such reaction mixtures. Isoasparagine, aspartic acid- β -hydroxamate, β -cyanoalanine and *threo*- β -hydroxyasparagine inhibited the transfer of ³H-AspNH₂ competitively. 5-Bromo-4-oxonorvaline at first was thought to inhibit transfer but the percentage inhibition increased with the age and the browning of the analogue solution, little inhibition was detected with freshly prepared solutions and so the inhibitory character of older solutions was attributed to decomposition products, including free bromine.

Substrate Specificity of Plants Producing β -Cyanoalanine

β -Cyanoalanine has been shown to occur in certain species of *Vicia*.¹⁶ For these experiments two *Vicia* species were selected, *V. faba*, which does not produce β -cyanoalanine and *V. sativa* whose seed contains β -cyanoalanine together with larger amounts of γ -glutamyl- β -cyanoalanine. The AspNH₂-tRNA synthetase from both seeds had very similar properties and both activated β -cyanoalanine in the exchange reaction with K_m values approximately 6×10^{-3} M (K_m for AspNH₂, 1.6×10^{-4} M) and V_{max} values of approximately 38%. β -Cyanoalanine (6×10^{-4} M) produced a 50% inhibition of ³H-AspNH₂ (at 10^{-5} M) transfer to tRNA. The possibility that β -cyanoalanine was being hydrolyzed to AspNH₂ either enzymically or chemically during the reaction period was considered, but the substrate potential of β -cyanoalanine did not increase when it was incubated for various periods up to 60 min with synthetase preparations prior to the ATP-PP_i exchange assay.

TABLE 3 THE EFFECT OF AspNH₂ ANALOGUES ON THE TRANSFER OF ³H-AspNH₂ TO tRNA USING ENZYME FROM *Phaseolus aureus*

Substrate	Concn required for 50% inhibition (M)	Substrate	Concn required for 50% inhibition (M)
Isoasparagine	7×10^{-5}	β -Cyanoalanine	9×10^{-4}
Aspartate- β -hydroxamate	5×10^{-4}	<i>threo</i> - β -Hydroxyasparagine	7×10^{-3}

³H-AspNH₂ present at a concentration of 10^{-5} M

*N*⁴-methylAspNH₂, α , γ -diaminobutyric acid, *S*-carbamoylcysteine, at concentrations of 10^{-2} M, inhibited transfer of ³H-AspNH₂ by 10–15%.

2-Amino-2-carboxyethanesulphonamide, *erythro*- β -hydroxyAspNH₂, *O*-phosphoserine, Aspdiamide, *meso*-diaminosuccinic acid monoamide, 5-diazo-4-oxonorvaline, 5-bromo-4-oxonorvaline, 5-chloro-4-oxonorvaline, β -methylAspNH₂, Asp- β -hydrazide and *N*- α -methylAspNH₂ failed to inhibit transfer when present at 10^{-2} M concentration.

Heterologous Acylation

P. aureus seed enzyme was able to acylate *t*RNA from *V. sativa* or *V. faba* with an efficiency of 83% of that determined for the homologous combination with *P. aureus t*RNA. *Vicia sativa* and *V. faba* enzymes were able to acylate *P. aureus t*RNA 68% as efficiently as the *P. aureus* enzyme. The *t*RNA of *V. sativa* and *V. faba* appeared to be completely interchangeable, acylation remaining constant when either heterologous combination was tested.

Preparations of enzyme and *t*RNA were obtained from the following tissues of *P. aureus*: (a) dry seed, (b) 5-day-old hypocotyl supernatant fraction, (c) 5-day-old hypocotyl, mitochondrial fraction, and (d) 14-day-old leaf chloroplast fraction. No cross-reaction was detected between fractions originating from seed and from mitochondria or chloroplasts. The only heterologous acylation reaction established was between seed and hypocotyl supernatant fractions. The seed enzyme could catalyze 20% transfer of ^3H -AspNH₂ to supernatant *t*RNA, whilst the supernatant enzyme could catalyze a 28% transfer to seed *t*RNA (percentages are relative to the activity measured with the respective homologous combinations).

Aspartyl-*t*RNA Synthetase

Enzyme Fractionation

Very low level activities of Asp-*t*RNA synthetase were detected in extracts of *P. aureus* seed after passage through Sephadex G25. However, a relatively late precipitate obtained during acetone fractionation (at 42–50%, v/v) exhibited high Asp-dependent ATP-PP_i exchange. This precipitate was either: (a) redissolved in dilute Tris-chloride buffer (0.03 M) and dialyzed against further buffer containing 15% (w/v) glycerol and 25 mM mercaptoethanol, when much inactive protein precipitated and was removed by centrifugation, or (b) redissolved in extraction buffer and subjected to ammonium sulphate fractionation, when a 45–55% saturated fraction was taken and exhaustively dialyzed against 0.03 M Tris-chloride. After either of these treatments, the enzyme was purified further on a DEAE-cellulose column. The enzyme eluted as a single peak between 85 and 105 ml of elution buffer, the full purification scheme is shown in Table 4.

TABLE 4. PURIFICATION OF ASP-*t*RNA SYNTHETASE FROM *Phaseolus aureus* SEED

Purification treatment	Total protein (mg)	Specific activity	Relative purification	Contaminating synthetases†
Homogenate‡	1300	1.9	1	5750
Acetone fraction	122	18.1	9.5	820
Ammonium sulphate fraction	46	39.2	20.6	260
DEAE-cellulose	4.2	326	171	11

* Specific activities are expressed as nmol PP_i exchanged/min/mg protein.

† Expressed as a % of Asp-*t*RNA synthetase in each fraction.

‡ Determined after removal of endogenous amino acids by Sephadex G25.

Optimum Assay Conditions

ATP-PP_i exchange The K_m for Asp was found to be 4.6×10^{-3} M, with 4×10^{-2} M required to obtain the V_{max} . The K_m for Mg^{2+} was 6.9×10^{-3} M with an optimum concentration of 1.2×10^{-2} M. The optimum pH was 8.2 with a sharp loss of activity below 7.8, but a plateau of high activity was maintained up to a pH of 9.1.

Maximum levels of activity were detected by incubating for 15 min at 37° with Tris-chloride buffer. Mn^{2+} and Co^{2+} were able to stimulate ATP-PP_i exchange, at rates of 35 and 12% of that determined with the optimal concentrations of Mg^{2+} .

Aminoacylation of tRNA Although the 40–50% (v/v) acetone fraction gave high values of Asp-dependent ATP-PP_i exchange, the fraction even after passage through Sephadex G75 to remove ribonucleases, gave very low levels of transfer of ³H-Asp to tRNA, only the DEAE-cellulose eluate catalyzed a significant reaction. The optimum tRNA concentration was 0.1 mg/ml, and concentrations higher than 0.3 mg/ml were inhibitory. The total level of aminoacylation reached a maximum after 10 min incubation at 37° and began to decrease after 15 min assay time. The ATP concentration was maintained at 10^{-3} M and the optimum Mg^{2+} concentration was 5×10^{-3} M with a K_m of 1.1×10^{-3} M. Mg^{2+} could be replaced by Mn^{2+} and Co^{2+} at 10^{-2} M, with retention of 56 and 34% activity, respectively. Spermine, spermidine and putrescine were able to substitute for Mg^{2+} with efficiencies of 32, 16 and 27% respectively.

Enzyme Stability

The enzyme could be stored in a dry form (e.g. as an acetone precipitate) for 1 month without loss of activity. However, in solution even in the presence of 40% (w/v) glycerol, and reagents containing stabilizing –SH groups, 50% of the activity was lost in 72 hr at –20°. A loss of 50% activity occurred when the enzyme was heated at 50° for 7 min, the half-life of the enzyme at 50° in the presence of ATP (4 μmol) alone and of ATP (4 μmol) and Asp (100 μmol) together, was increased to 12 and 23 min, respectively.

Concentrations of urea above 4 M caused enzyme denaturation in 5 min at 37°, and only 30% of the activity was recovered after dialysis. *p*-CMB inhibited the enzyme and a K_i was determined for the DEAE-cellulose purified enzyme (protein concentration 30 μg/ml) of 4×10^{-5} M. Sodium azide (10^{-2} M) did not inhibit the enzyme.

Salt Effects

The enzyme, when assayed by ATP-PP_i exchange, was inhibited by most monovalent ions. Concentrations of Li^+ , NH_4^+ , Na^+ , K^+ and Cs^+ (as chlorides) producing 50% inhibition were 0.6, 0.9, 1.4, 2.3 and 1.1 M respectively, the similar concentrations for IO_3^- , ClO_3^- , CNS^- , I^- , Br^- , Cl^- , NO_3^- and $MeCOO^-$, were 0.04, 0.1, 0.1, 0.1, 0.7, 2.2, 0.7 and 2.0 M, respectively. The aminoacylation of tRNA was very sensitive to the addition of ions, 50% inhibition values for Li^+ , NH_4^+ , Na^+ , K^+ and Cs^+ were determined as 0.11, 0.14, 0.18, 0.22 and 0.16 M, respectively.

Amino Acid Substrate Specificity

ATP-PP_i exchange reaction. The ability of Asp-tRNA from *P. aureus* to utilize various analogues is shown in Table 5. Lineweaver–Burk plots for the substrates were linear. No inhibition of Asp-dependent ATP-PP_i exchange was detected for the analogues tested.

Aminoacyl transfer to tRNA The possibility of transfer of an analogue to *tRNA* was measured indirectly (cf AspNH₂-synthetase) by studying the inhibition of transfer of ³H-Asp caused by various concentrations of analogues (Table 6)

TABLE 5 KINETIC PARAMETERS OF THE ATP-PP_i EXCHANGE STIMULATED BY ASP AND SEVERAL ANALOGUES USING Asp-*tRNA* SYNTHETASE FROM *Phaseolus aureus* SEED

Substrate	V_{\max}	Concn for V_{\max} (M)	K_m
Aspartic acid	100	4×10^{-2}	4.6×10^{-3}
α -Aminomalonic acid	105	8×10^{-2}	7.2×10^{-3}
<i>threo</i> - β -Hydroxyaspartic acid	78	10^{-1}	2.4×10^{-2}

Results are expressed with respect to the L-form of the amino acid, K_m values are expressed as molar concentrations, V_{\max} expressed as a % of the V_{\max} for Asp *threo*- β -MethylAsp, *erythro*- β -hydroxyAsp, *O*-phosphoserine, *O*-methylserine, homoserine, β -cyanoalanine, isoasparagine, *N*-acetylAsp, 5-bromo-4-oxonorvaline and 5-chloro-4-oxonorvaline failed to stimulate ATP-PP_i exchange at 5×10^{-2} M

The Dependence of Asp-Stimulated ATP-PP_i Exchange upon Asp-tRNA Synthetase

The low transfer of ³H-Asp to *tRNA* effected by an enzyme fraction stimulating a high Asp-dependent ATP-PP_i exchange might indicate that the preparation contained more than one enzyme able to catalyze ATP-PP_i exchange. Asparagine synthetase and arginosuccinate synthetase are other enzymes that are able to catalyze this reaction. However, when purified Asp-*tRNA* synthetase was incubated with ³H-Asp, ATP, Mg²⁺ and (a) ammonia, (b) glutamine, or (c) citrulline, neither ³H-AspNH₂ from (a) or (b) nor ³H-arginosuccinate (from c) were detected after PC separations.

TABLE 6 THE EFFECT OF ASP ANALOGUES ON THE TRANSFER OF ³H-Asp TO *tRNA* USING ENZYME FROM *Phaseolus aureus*

Substrate	Concn required for 50% inhibition (M)	Substrate	Concn required for 50% inhibition (M)
α -Aminomalonic acid	6×10^{-5}	<i>threo</i> - β -Methylaspartic acid	4×10^{-3}
<i>threo</i> - β -Hydroxyaspartic acid	8×10^{-4}	<i>erythro</i> - β -Hydroxyaspartic acid	1×10^{-2}

³H-Asp present at a concentration of 10^{-5} M

O-phosphoserine, β -cyanoalanine, homoserine, *O*-methylserine, *N*-acetylAsp, 5-bromo-4-oxonorvaline and 5-chloro-4-oxonorvaline failed to inhibit transfer of ³H-Asp to *tRNA*

Glutaminyl-tRNA Synthetase

Enzyme Fractionation

Low or zero activities assayed by ATP-PP_i exchange were determined in crude preparations of *P. aureus* seed after passage through Sephadex G25. However, the formation of ¹⁴C-GluNH₂-*tRNA* was catalyzed by crude preparations, after passage through Sephadex

G75 to remove ribonucleases Active preparations of GluNH₂-*t*RNA synthetase were obtained by employing the following fractionation procedure (a) precipitate as 50–60% saturated ammonium sulphate fraction, (b) precipitations between 40 and 50% (v/v) acetone, and (c) passage through Sephadex G75 to remove low MW compounds Enzyme was also prepared from cotyledons of *A. julibrissin* seedlings (radicles 1 cm in length) by the same purification procedure as that described for *P. aureus* (Table 7)

TABLE 7 PURIFICATION OF GluNH₂-*t*RNA SYNTHETASE FROM *Phaseolus aureus* AND *Albizzia julibrissin*

Plant	Purification treatment	Total protein (mg)	Specific activity*	Relative purification
<i>P. aureus</i>	Homogenate†	1300	2.1	1
	Ammonium sulphate fraction	226	10.8	5.1
	Acetone fraction	32	62.4	29.6
<i>A. julibrissin</i>	Homogenate†	946	1.3	1
	Ammonium sulphate fraction	152	8.1	6.2
	Acetone fraction	26	45.1	34.7

* Specific activity expressed as pmol ¹⁴C-GluNH₂ transferred to *t*RNA/min/mg protein

† Determined after removal of endogenous amino acids and RNAase by Sephadex G75

Evidence for the Formation of ¹⁴C-GluNH₂-*t*RNA

Wilcox and Nirenberg¹⁴ (see Introduction) have indicated that GluNH₂ is not transferred directly to *t*RNA. It was therefore important to show in these experiments that the GluNH₂ was transferred directly and that, GluNH₂-*t*RNA was not derived by amidation of Glu-*t*RNA. The transfer of ¹⁴C-GluNH₂ (10⁻⁵ M) was 50% inhibited by the addition of 10⁻⁵ M ¹²C-GluNH₂ when *t*RNA was limiting. In contrast the formation of GluNH₂-*t*RNA was not inhibited by the same concentration of Glu, although slight inhibition was produced by concentrations of 10⁻³ M. No Glu-*t*RNA synthetase activity could be detected in the enzyme preparation used but it catalyzed some conversion of ¹⁴C-Glu to ¹⁴C-GluNH₂, the conversion of ¹⁴C-GluNH₂ to ¹⁴C-Glu did not occur under the same conditions. These observations are consistent with the direct transfer of GluNH₂ to *t*RNA.

Optimum Assay Conditions

ATP-PP_i exchange No GluNH₂-dependent ATP-PP_i activity could be detected with the purified extracts from *P. aureus* (although slight exchange was detected with cruder preparations). However, there was evidence of the stimulation of a GluNH₂-dependent ATP-PP_i exchange by the addition of *t*RNA at pH 7.4. This stimulation (which was complicated by the adsorption of ³²PP_i on to the *t*RNA, giving higher endogenous counts) never amounted to more than a 200% increase above the endogenous level of ATP-PP_i exchange, which was insufficient for accurate study of the enzyme's properties.

Aminoacylation of *t*RNA A concentration of 4 × 10⁻⁶ M (0.2 μCi) GluNH₂ was routinely used for all assays. The optimum *t*RNA concentration was 0.15 mg/ml; concentrations higher than 0.6 mg/ml were inhibitory. The total level of aminoacylation reached a maximum after 8 min incubation at 37°, and began to decrease after 12 min. ATP was present at 10⁻³ M, and the optimum Mg²⁺ concentration was 2.5 × 10⁻² M (*K_m* 8.9 × 10⁻³ M). The pH

activity curve showed a sharp peak at 7.8. Only Mn^{2+} among the divalent cations tested could replace Mg^{2+} at 10^{-1} M (42% of maximum activity with Mg^{2+}). Spermine, putrescine and spermidine (all at 10^{-1} M) could also partially substitute for Mg^{2+} (comparable activities were 35, 32 and 12%, respectively). The transfer of ^{14}C -GluNH₂ to tRNA was inhibited by salts, the concentrations required for 50% inhibition by sodium salts of Cl^- , Br^- and I^- were 0.2, 0.1 and 0.06 M respectively.

TABLE 8 THE EFFECT OF GluNH₂ ANALOGUES ON THE TRANSFER OF ^{14}C -GluNH₂ TO tRNA USING ENZYME FROM *Phaseolus aureus* AND *Albizia julibrissin*

Substrate	Concn required for 50% inhibition (M)	
	<i>P. aureus</i>	<i>A. julibrissin</i>
γ -Methyleneglutamine	7.1×10^{-3}	1.6×10^{-2}
Theanine	5.6×10^{-3}	1.2×10^{-3}
S-Carbamoylcysteine	2.2×10^{-3}	6.4×10^{-3}
O-Carbamoylserine	1.1×10^{-3}	3.1×10^{-3}
Albizzine	8.1×10^{-4}	9.2×10^{-4}
Glutamic acid γ -hydroxamate	3.8×10^{-2}	4.1×10^{-2}

^{14}C -GluNH₂ present at a concentration of 4×10^{-6} M

Glutamic acid diamide gave 2–5% inhibition at 4×10^{-2} M for both enzymes

Amino Acid Substrate Specificity

The low rates of ATP-PP_i exchange activity prevented the use of this technique in studies on amino acid substrate specificity. The possibility that several analogues of GluNH₂ could be utilized by the enzyme was therefore investigated by the indirect method, i.e. by measuring the inhibitory action upon the transfer of ^{14}C -GluNH₂ to tRNA. The concentrations of the analogues required to produce 50% inhibition of transfer in systems employing enzymes from *P. aureus* and *A. julibrissin* are shown in Table 8.

DISCUSSION

AspNH₂-tRNA Synthetase

The level of contaminating synthetases (assayed by ATP-PP_i exchange) was reduced to a level ($\approx 20\%$) that would not interfere with substrate specificity studies carried out by the exchange assay. The transfer of ^3H -AspNH₂ to tRNA was not inhibited by the addition of a mixture containing the other normal 19 amino acid constituents of protein, and is therefore presumed to be catalyzed by one specific enzyme. The enzyme was stable in the cold, but rather readily deactivated by heating, and partially protected against heat denaturation by ATP and AspNH₂, the protection of aminoacyl-tRNA synthetases against heat has been discussed by Norris and Fowden^{18,19}. The enzyme was denatured by 4 M urea indicating that hydrogen bonding plays an essential part in retaining the enzyme's conformation, and also by low concentrations of *p*-CMB indicating that -SH groups are present at or near the active site of the enzyme. The ability of Mn^{2+} , Co^{2+} or Ni^{2+} to replace Mg^{2+} in the

¹⁸ NORRIS, R. D. and FOWDEN, L. (1972) *Phytochemistry* **11**, 2921

¹⁹ NORRIS, R. D. and FOWDEN, L. (1973) *Biochim. Biophys. Acta* in press

aminoacylation reaction has been well documented²⁰ However, the failure of the three polyamines tested to substitute partially for Mg^{2+} is unusual

The hierarchy of salt inhibition for the ATP-PP_i exchange reaction, $Li^+ > NH_4^+ > Cs^+ > Na^+ > K^+$ and $I^- > Br^- > Cl^-$ is similar to that shown for the Pro-*t*RNA synthetase from *P. aureus*¹⁸ The resistance of the ATP-PP_i reaction to high salt concentration indicates that the binding of the ATP and amino acid to the enzyme is not by ionic forces The sensitivity of the aminoacylation reaction to salts has frequently been reported in animals and micro-organisms,²¹ and may be due to (a) binding of *t*RNA to the enzyme by ionic bonds, or (b) conformational changes in the structure of the *t*RNA molecule The fit of the data to the Debye-Huckel equation would indicate that *t*RNA is bound to the enzyme by simple interacting charges However, the Debye-Huckel theory is only applicable to ionic strengths up to 0.1 M, and so the slopes of this type of graph should be used with caution to estimate the number of charges involved in this type of reaction²¹ It has been suggested that the large differences in aminoacylation rate occurring over a narrow range of salt concentrations could form the basis of an *in vivo* mechanism for controlling protein synthesis²²

The ability of low concentrations of various ions to stimulate the aminoacylation of *t*RNA has been reported previously,^{22,23} but the large increases in aminoacylation caused by organic solvents in other systems²⁴ could not be detected with the AspNH₂ enzyme

The most interesting result in substrate specificity studies was the lack of inhibition by aspartic acid diamide The amide is usually a potent inhibitor of aminoacyl-*t*RNA synthetases with a K_i value close to the K_m of the corresponding amino acid^{18,26} This result would indicate that the carboxyl group of the amino acid is essential for binding to the enzyme Presumably when isoasparagine acts as a substrate, the molecule is positioned in such a manner that the β -carboxyl group and α -carbamoyl group are able to bind to the active site in a fashion similar to that of AspNH₂ Asp- β -hydroxamate was found to stimulate ATP-PP_i exchange in *Lactobacillus*,²⁷ and so the N^4 -hydroxyl group clearly does not prevent binding to the enzyme The ability of the *threo* isomer (in contrast to the *erythro* form) of β -hydroxy-AspNH₂ to act as an inhibitor is probably due to steric hindrance of the binding of the α -amino group by the *threo* hydroxyl None of the derivatives of 4-oxonorvaline, which are potent inhibitors of AspNH₂ synthesis in tumour cells,²⁸ acted as substrates or inhibitors of the enzyme, clearly being too large to be accommodated at the active site of the enzyme

An investigation into the properties and substrate specificity of the enzyme from 2 species of *Vicia* showed that no significant differences occurred between β -cyanoalanine producing and non-producing species As no direct incorporation into *t*RNA has been measured it is possible that the analogue only inhibits the formation of AspNH₂-*t*RNA and is not itself incorporated into protein Furthermore the endogenous concentration of β -cyanoalanine in the seed of *V. sativa* may be insufficient in comparison with that of AspNH₂ (which has a K_m value 30 times smaller than β -cyanoalanine) to effect any significant competition,

²⁰ MEHLER, A. H. and CHAKRABURTY, K. (1972) in *Advances in Enzymology* (MEISTER, A., ed.), Vol. 35, p. 443, Interscience, New York

²¹ LOFTFIELD, R. B. (1971) in *Protein Synthesis* (MC CONKEY, E., ed.), Vol. 1, p. 1, Marcel Dekker, New York

²² SMITH, D. W. E. (1969) *J. Biol. Chem.* **244**, 896

²³ YU, C. T. and HIRSH, D. (1967) *Biochim. Biophys. Acta* **142**, 149

²⁴ LOFTFIELD, R. B. and EIGNER, E. A. (1967) *J. Biol. Chem.* **242**, 5355

²⁵ RITTER, P. O., HULL, F. J. and JACKOBSON, K. B. (1969) *Biochim. Biophys. Acta* **179**, 524

²⁶ PAPAS, T. S. and MEHLER, A. H. (1970) *J. Biol. Chem.* **245**, 1588

²⁷ NORTON, S. J. and CHEN, Y. T. (1969) *Arch. Biochem. Biophys.* **129**, 560

²⁸ CHOU, T. C. and HANDSCHUMACHER, R. E. (1972) *Biochem. Pharmacol.* **21**, 39

either as a substrate or as an inhibitor. If this were so, then no evolutionary pressure would exist in *V. sativa* leading to a modification of the substrate specificity of its AspNH₂-tRNA synthetase. It is also intriguing to realize that even if β -cyanoalanine was incorporated into the plant's proteins, such residues might be converted into AspNH₂ moieties by subsequent nitrilase action.

There have been conflicting reports concerning the ability of a number of cytoplasmic, mitochondrial and chloroplastic synthetases to cross-react with their heterologous tRNAs.¹⁻²⁹ A further more detailed study into the multiplicity of tRNA^{AspNH₂} in plants will be carried out.

Aspartyl-tRNA Synthetase

The degree of purity of this synthetase obtained from *P. aureus* was similar to that of the AspNH₂ enzyme. The idea that Asp-dependent ATP-PP_i exchange catalyzed by the enzyme preparation was due to contaminating AspNH₂ synthetase or arginosuccinate synthetase activity was disproved, and no AspNH₂- or Glu-dependent ATP-PP_i exchange could be detected in the purified enzyme fraction. The transfer of ³H-Asp to tRNA was not inhibited by a mixture of the other 19 protein amino acids. The low level of aminoacylation observed may be due to the presence of an inhibitor in the enzyme and/or tRNA preparations or possibly to a low level of tRNA^{Asp} in the extracted tRNA.

The effect of 4 M urea, *p*-CMB and sodium azide were similar to that demonstrated for the AspNH₂ enzyme. Mn²⁺ and Co²⁺ were again able to substitute for Mg²⁺, but with this enzyme the three polyamines were also able to substitute (cf. results with *E. coli* aminoacyl-tRNA synthetases).³⁰

The ATP-PP_i exchange reaction was a little more sensitive to salt than that catalyzed by the AspNH₂ enzyme, however, the degree of salt inhibition was not comparable with the extreme sensitivity by Glu-tRNA synthetase from *P. aureus*.³¹ The aminoacylation reaction showed the same high sensitivity to salts as that described for the AspNH₂ enzyme.

α -Aminomalonate has been shown to act as an Asp analogue in the following enzyme systems: (a) Glu-dependent AspNH₂ synthetase,³² (b) Asp-decarboxylase,³³ and (c) Asp-transcarbamylase.³⁴ The ability of the Asp-tRNA synthetase to recognize this lower homologue of Asp has parallels in the binding of smaller analogue molecules to other plant aminoacyl-tRNA synthetases, e.g. 2-amino-4-methylhex-4-enoic acid by Phe-tRNA synthetase³⁵ and azetidine-2-carboxylic acid by Pro-tRNA synthetase.¹⁸ The observation that the *threo* stereoisomer, but not the *erythro* form of β -hydroxyAsp acts as an analogue for Asp is in agreement with observations with other enzymes, e.g. arginosuccinate synthetase,³⁶ Asp-decarboxylase,³³ and GluNH₂-dependent AspNH₂ synthetase.³² However, the stereospecificity is reversed when β -hydroxyAsp isomers are tested with Asp-transcarbamylase³⁴ and Asp-aminotransferase.³⁷ Irrespective of the stereoisomer of β -hydroxyAsp utilized by an enzyme the corresponding β -methyl form is usually a poorer substrate,³²⁻³⁴ although this is not so for arginosuccinate synthetase.³⁶ It is possible that with the enzyme under study the

²⁹ BURKARD, G., GUILLEMAUT, P. and WEIL, J. H. (1970) *Biochim. Biophys. Acta* **224**, 184.

³⁰ TAKEDA, Y. and IGARASHI, K. (1969) *Biochem. Biophys. Res. Commun.* **37**, 917.

³¹ LEA, P. J. and FOWDEN, L. (1972) *Phytochemistry* **11**, 2129.

³² HOROWITZ, B. and MEISTER, A. (1972) *J. Biol. Chem.* **247**, 6708.

³³ TATE, S. S. and MEISTER, A. (1970) in *Advances in Enzymology* (MEISTER, A., ed.), Vol. 35, p. 503, Interscience, New York.

³⁴ DAVIES, G. E., VANAMAN, T. C. and STARK, G. (1970) *J. Biol. Chem.* **245**, 1175.

³⁵ ANDERSON, J. W. and FOWDEN, L. (1970) *Biochem. J.* **116**, 677.

³⁶ RACHOVANSKY, O. and RATNER, S. (1968) *Arch. Biochem. Biophys.* **127**, 688.

³⁷ JENKINS, W. T. (1961) *J. Biol. Chem.* **236**, 1121.

erythro-hydroxyl group would interfere with the attachment of the α -amino group to the active site, the *threo* group being on the other side of the molecule would not exert such an effect

Glutamyl-tRNA Synthetase

The lack of ATP-PP_i exchange catalyzed by GluNH₂-*t*RNA synthetase in the absence of *t*RNA is in agreement with the properties reported for this enzyme from other sources¹¹⁻¹³ The low GluNH₂-dependent ATP-PP_i exchange demonstrated in the presence of *t*RNA could be due to an inhibitor in the *t*RNA preparation or to the lack of an active specific *t*RNA^{GluNH₂}, *t*RNA probably behaves as an activator for the enzyme in the ATP-PP_i exchange reaction and not as an acceptor of the aminoacyl group³⁸ However, the *P. aureus* enzyme was able to catalyze the formation of ¹⁴C-GluNH₂-*t*RNA; this reaction was inhibited by the addition of unlabelled GluNH₂ or of high concentrations of Glu The enzyme probably contained low glutamine synthetase activity producing small amounts of inhibitory GluNH₂ when Glu was added at high concentrations

The optimum Mg²⁺ ATP ratio for the aminoacylation of *t*RNA was 25:1 for the GluNH₂ enzyme, 12.5:1 for the AspNH₂ enzyme, and 5:1 for the Asp enzyme It is clear that the Mg²⁺ ATP ratio must be varied during initial experiments designed to determine the optimum conditions for a particular aminoacylation reaction

The enzyme was more specific than either the AspNH₂ or the Asp enzyme in its use of divalent cations in place of Mg²⁺, but was able to utilize all three polyamines The extent of inhibition of aminoacylation by low concentrations of salts was similar to that shown by the other two enzymes, and so again the activity of an enzyme concerned in the initial step in protein synthesis was quite susceptible to small changes in salt concentrations

None of the analogues tested acted as potent inhibitors of the transfer of ¹⁴C-GluNH₂ to *t*RNA; for instance at least a 200-fold excess of albizzine was required to produce a 50% inhibition of aminoacylation There was little difference between the enzymes isolated from *P. aureus* and *A. julibrissin* (a natural producer of albizzine), although the *A. julibrissin* enzyme required slightly higher concentrations of analogues to produce 50% inhibition

Although the seed of *A. julibrissin* contains a substantial concentration of albizzine, the transfer of significant amounts of the analogue to RNA or into protein seems unlikely, and probably it has little effect upon the formation of GluNH₂-*t*RNA It is possible that within the cell there is compartmentation which prevents the access of albizzine to the ribosomes

The negligible inhibition by Glu-diamide again indicates the possible involvement of the α -carboxyl group in the binding of the amino acid to the enzyme The presence of a C-N-C bond (albizzine), a C-O-C bond (*O*-carbamoylserine), or a C-S-C bond (*S*-carbamoylcysteine), instead of the C-C-C bond of GluNH₂ alters the distance between the β -carbon and the amide carbon atom from 3.08 Å in GluNH₂, to 2.94, 2.86 and 3.62 Å, respectively, in three analogues³⁹ The inhibitory action of these analogues (albizzine > *O*-carbamoylserine > *S*-carbamoylcysteine) clearly reflects these differences in molecular size Similarly, structural alterations that increase the size of the amide group, e.g. the introduction of a hydroxamate or an *N*⁵-ethyl substituent (theanine), have a marked influence on enzyme binding The relative inertness of γ -methyleneglutamine as a GluNH₂ analogue may be compared with the inability of γ -methyleneglutamic acid to act as a substrate for the Glu-*t*RNA synthetase from *P. aureus*³¹

³⁸ MITRA, S. K. and MEHLER, A. H. (1967) *J. Biol. Chem.* **242**, 5490

³⁹ PAULING, L. (1948) in *The Nature of the Chemical Bond*, Cornell University Press, New York

EXPERIMENTAL

Plant materials Seeds of *A. julibrissin* were a gift from Dr M Mazelis, Davis, California, U S A. Seeds of *P. aureus*, *V. sativa* and *V. faba* were obtained commercially.

Amino acids All samples were tested for purity by PC. The L-isomers of the protein amino acids were high purity commercial samples. Other amino acids were obtained as follows: L-threo- β -MethylAsp was a gift from Dr H J Bright, Pennsylvania, U S A. DL-threo- β -HydroxyAsp was prepared by reducing the N-benzyl derivative, a gift from Dr C H Stammer, Georgia, U S A. DL-erythro- β -HydroxyAsp was a gift from Professor J G Morris, Aberystwyth. Meso-diaminosuccinic acid monoamide 2HBr, 5-diazo-4-oxo-L-norvaline, 5-bromo-4-oxo-L-norvaline, 5-chloro-4-oxo-L-norvaline HCl and DL- β -methylAspNH₂ were gifts from Professor R E Handschumacher, Connecticut, U S A. L-Asp- β -hydrazide, S-carbamoyl-L-cysteine and N- α -methyl-DL-AspNH₂ were gifts from Dr I Chibata, Tanabe Seiyaku Co Ltd, Osaka, Japan. L-erythro- β -HydroxyAspNH₂ and L-threo- β -hydroxyAspNH₂ were gifts from Dr A Singerman, Jerusalem. IsoAspNH₂ was a gift from Dr G L Tritsch, New York, U S A. L-Glu-diamide was a gift from Professor F C Steward, Cornell, U S A. Theanine was a gift from Dr K Sasaoka, Kyoto, Japan. L- γ -MethyleneGluNH₂ (*Arachis hypogea*) and L-albizzine (*Acacia armata*) were natural isolates. The potassium salt of α -aminomalonic acid and L-N-methylAspNH₂ were synthesized in our laboratory. DL-O-Phosphoserine (Koch-Light & Co Ltd), L-2-amino-2-carboxyethanesulphonamide (Ciba, New Jersey, U S A), L- β -cyanoalanine and L- α , γ -diaminobutyric acid (Calbiochem, U S A), DL-N-acetylAsp, DL- β -Asphydroxamate and L- γ -Gluhydroxamate (Sigma & Co Ltd), were obtained commercially.

Radioisotopic chemicals L-[G-³H]Asparagine (185 mCi/mmol), L-[G-³H]aspartic acid (220 mCi/mmol), L-[U-¹⁴C]glutamic acid (225 mCi/mmol), L-[U-¹⁴C]glutamine (50 mCi/mmol) and sodium [³²P]orthophosphate were obtained from the Radiochemical Centre, Amersham.

Enzyme fractionation All operations were carried out in a cold room at 1–4°. Seed meal was ground manually with extraction medium (0.1 M Tris-chloride buffer pH 7.8, containing 15% (w/v) glycerol, 20 mM MgCl₂ and 25 mM mercaptoethanol), 10 ml/g seed, and the extract subjected to acetone and ammonium sulphate fractionation as described in an earlier paper.³¹ The precise fractions taken are described in the text, for each of the enzymes under investigation. Sephadex G75 chromatography was carried out on columns equilibrated with the extraction medium at a strength of 0.05 M Tris DEAE-cellulose chromatography was carried out on a column (32 × 1.7 cm dia.) equilibrated with 0.03 M Tris-chloride buffer, pH 7.8, containing 15% (w/v) glycerol and 25 mM mercaptoethanol, and developed with a linear gradient of 0–0.4 M KCl (200 ml in each gradient vessel). Chloroplasts were isolated by the method described by Leech⁴⁰ and purified on a sucrose gradient. The enzymes were solubilized by stirring the chloroplasts in 0.1% Triton X100 for 30 min and centrifuging off the chloroplast membranes at 15 000 g for 15 min. Mitochondria were prepared by the method described by Bonner,⁴¹ and washed twice to remove soluble enzymes. Mitochondrial enzymes were solubilized in 0.1% Triton X100 followed by centrifugation at 30 000 g for 30 min.

tRNA Isolation The method of Vanderhoef *et al.*⁴² was employed to extract tRNA from the chloroplasts, mitochondria and supernatant extracts. Seed meal was extracted by the same method but double the suggested volume of Tris-chloride buffer was used in the initial extraction. Preparations had E_{260}/E_{280} ratios > 2.0 and E_{260}/E_{230} ratios > 1.8 before use in enzyme assays.

Protein determination The method of Lowry *et al.*⁴³ was normally adopted, using bovine serum albumin as a standard. Fractions eluted from DEAE-cellulose columns were assayed by determining absorbance at 260 and 280 nm by the method described by Layne.⁴⁴

Assay Procedures (a) *ATP-PPi exchange system* The method was based on that described by DeMoss and Novelli.⁴⁵ The precise assay mixtures are discussed for each individual enzyme in the text. The exchange rates were calculated by the method of Davie *et al.*⁴⁶ (b) *Aminoacyl-tRNA formation* The method followed was that described by Vanderhoef *et al.*,⁴² precipitates were collected on glass fibre discs and radioactivity determined by scintillation counting. Individual assay mixtures are discussed in the text.

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⁴⁰ LEECH, R. M. (1963) *Biochim Biophys Acta* **71**, 253.

⁴¹ BONNER, W. D. (1967) in *Methods in Enzymology* (ESTABROOK, R. W. and PULLMAN, M., eds), Vol. 10, p. 126, Academic Press, New York.

⁴² VANDERHOEF, L. N., BOHANNON, R. F. and KEY, J. L. (1970) *Phytochemistry* **9**, 2291.

⁴³ LOWRY, O. H., ROSEBROUGH, N. J., FARR, A. L. and RANDALL, R. J. (1951) *J. Biol. Chem.* **193**, 265.

⁴⁴ LAYNE, E. (1957) in *Methods in Enzymology* (COLOWICK, S. P. and KAPLAN, N. O., eds), Vol. 3, p. 447, Academic Press, New York.

⁴⁵ DEMOSS, J. A. and NOVELLI, G. D. (1956) *Biochim Biophys Acta* **22**, 49.

⁴⁶ DAVIE, E. W., KONINGSBERGER, V. V. and LIPMANN, F. (1956) *Arch. Biochem. Biophys.* **65**, 21.